## Chemistry - Mrs. Bauck, PHUHS

Unit 10: Thermochemistry - Chapter 14 (Chem 1H)

State Standards (\*\*\* = Chem 1H only)

## **Topic 1: Forms and Transformations of Energy**

SC.912.P.10.1 Differentiate among the various forms of energy and recognize that they can be transformed from one form to others.

## **Topic 2: Endothermic and Exothermic Reactions**

SC.912.P.10.6 Create and interpret potential energy diagrams, for example: chemical reactions, orbits around a central body, motion of a pendulum.

SC.912.P.10.7 Distinguish between endothermic and exothermic chemical processes.

\*\*\*SC.912.P.10.8 Explain entropy's role in determining the efficiency of processes that convert energy to work.

	Students will be able to:			
4.0		☐ Describe how a fusion nuclear		
4.0	Extensions/Applications	Explain how Tokamak magnet		
		☐ Solve advanced calorimetry pr	oblems.	
		ents will be able to:		
			in all physical and shamical processes	
			in all physical and chemical processes.	
		exists in discrete amounts.	nuclear levels energy is not continuous but	
			our valete d there als Einsteinle acception	
		Explain now energy and mass $E=mc^2$ .	are related through Einstein's equation.	
		<ul> <li>Describe how properties of ato</li> </ul>	mic nuclei are responsible for energy-	
		related phenomena such as rad	ioactivity, fission and fusion.	
			ropy and energy that accompany chemical	
		reactions influence reaction pa		
		☐ Understand that chemical react	ions result in the release or absorption of	
		energy.		
3.0	Learning Goal (Derived from State Standard)	☐ Compare and contrast endother		
			lectromagnetism explains that electricity	
			ted. Electric charges are the source of	
		electric fields. Moving charges		
			the propagation of a disturbance. They	
		transport energy and momentu		
		☐ Solve specific heat problems, t		
			combustion, formation. (chemical changes)	
			porization, solidification, condensation,	
		solution. (physical changes)		
		☐ Explain Hess' Law and work c		
		1	equation and work corresponding	
		problems.		

2.0	Required Skills or Background Knowledge to accomplish Learning Goal	Students will be able to:  Summarize the Law of Conservation of Energy and give examples.  Explain how energy is quantized in an atom.  Define radioactivity.	
		<ul> <li>□ Compare and contrast fission and fusion.</li> <li>□ Give a basic definition of endothermic and exothermic reactions.</li> <li>□ Define kinetic energy and potential energy.</li> <li>□ Write and balance chemical equations.</li> </ul>	
1.0	With help from the teacher, student has partial success with the goal	With help from a teacher, students will be able to:  ☐ Achieve partial success with 2.0 and/or 3.0.	
0.0	Even with help, the student has no success with the goal	□ Even with help, student is unable to understand or complete any of the skills in scales 1.0 through 4.0.	